

Name of Institute: Indus Institute of Sceinces, Humanities and Liberal Studies Name of Faculty : Department of Languages (English)

Course code: EN0111

Course name: Technical Communication Pre-requisites: Basic English Vocabulary & Grammar Credit points: 03 Offered Semester: 01

Course Coordinator (weeks 12)

Full name: Dr. Mamta Pillai (HoD, Dept. of Languages, IISHLS, IU) Department with siting location: Computer Lab (Grd. Flr.) (MB) Telephone: 9924241816 Email: <u>mamtapillai.gd@indusuni.ac.in</u> Consultation times: 4:00 to 5:00PM from Monday to Friday

Course Lecturer (weeks 12)

Full name: Ms. Pranjal Bhatt/ Ms. Foram Patel/Ms. Nidhi Singh Department with siting location: Equinox Lab, Grd Floor (MB) Telephone: 9429296329/9537384053/8160976525 Email: pranjalbhatt.gd@indusuni.ac.in/forampatel.ishls@indusuni.ac.in/nidhisingh.ishls@indusuni .ac.in

Consultation times: 4 to 5PM from Monday to Friday

Students will be contacted throughout the session via mail with important information relating to this course.

Course Objectives

By participating in and understanding all facets of this course a student will:

- 1) To enable students to do away with their inhibitions and stage-fright
- 2) To enhance their basic vocabularyand knowledge of sentence structure through Activities
- 3) To help them learn non verbal communication
- 4) To enable technology aided language learning

Course Outcomes (CO)

- 1) To help students develop comprehension and soft skills
- 2) To increase student's ability to improve and utilize the technical skills necessary for reading and writing.
- 3) To improve students' communication skills in both technical and professional contexts.
- 4) To help students improve the articulation and express thoughts fluenty in the second language.

Course Outline

- 1) Listening Skills
- 2) Public Speaking activities
- 3) Grammar & Vocabulary
- 4) Reading and Comprehension

Course Content

Unit 1: Listening

- Icebreaking Switch Introduction
- Icebreaking Past, Present and Future
- Listening/ Cloze Test 1
- Listening/ Ted Talk
- Listening/ Josh Talk
- Listening/Celebrity Interviews
- Listening/ News Hour debates

Unit 2: Speaking

- Introduction and Polite Conversations
- Situational Dialogues
- Role Play
- Body Language
- Group Discussion
- JAM Sessions

Unit 3: Language Focus

- Teaching Remedial Grammar through Poem
- Vocabulary Building/Describing Words
- Movie Viewing and Discussion
- Book and Movie Adaptation
- Match Commentary and Review

Unit 4: Reading Skills

- Reading/Reading with Vocalic
- Read and Tweet
- Skimming/Scanning the Newspapers
- Reading/Play Enactment
- Reading/Short Stories

Method of delivery

- 1) Communicative Language Teaching (Learner Centric)
- 2) Face to face lectures
- 3) Task Based Language Learning
- 4) The Lexical Approach

Study time

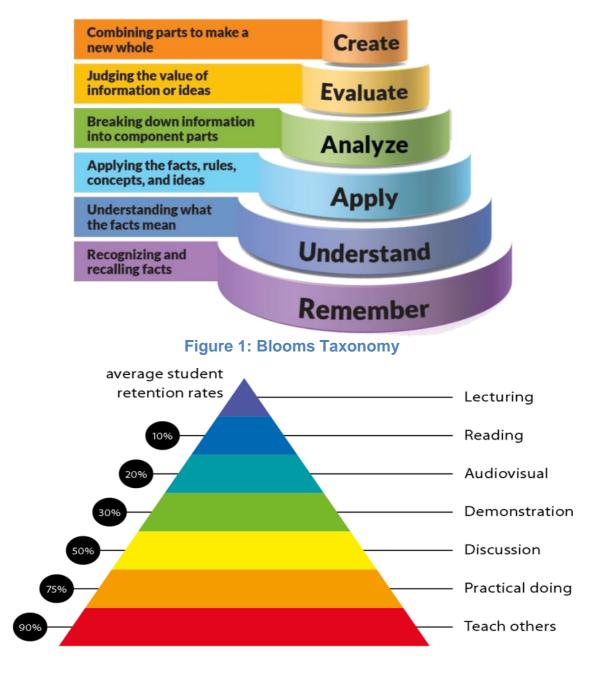
03 Hours



	P01	PO2	PO3	PO4	PO5	P06	PO7	PO8	PO9	PO10	P011	PO12
CO1	-	1	1	1	2	1	-	1	2	3	-	2
CO2	-	2	1	1	1	1	-	1	3	3	1	2
CO3	-	3	1	1	2	1	-	2	3	3	1	2
CO4	-	2	1	1	1	1	-	1	3	3	1	2

CO-PO Mapping (PO: Program Outcomes)

Blooms Taxonomyand Knowledge retention(For reference) (Blooms taxonomy has been given for reference)







Graduate Qualities and Capabilities covered (Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department ofGraduate Capabilities
Informed Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area of study has developed and how it relates to other areas.	1 Professional knowledge, grounding & awareness
Independent learners Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate information, using a variety of sources and technologies. Acknowledge the work and ideas of others.	2 Information literacy, gathering & processing
Problem solvers Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	4 Problem solving skills
Effective communicators	5 Written communication
Articulate ideas and convey them	6 Oral communication
effectively using a range of media. Work collaboratively and engage with people in different settings. Recognize how culture can shape communication.	7 Teamwork
Responsible Understand how decisions can affect others and make ethically informed choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	10 Sustainability, societal & environmental impact



Lecture/Tutorial Time: (Give lecture times in the format below)

(1Hour theory and 2Hours practical per week) As per the Master Time-Table of FY B.Tech Year -2021

Attendance Requirements

The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for semester examinations.

Details of referencing system to be used in written work: NIL

Text books: NIL

References:

- English for Engineers and Technologists, Volumes 1 and 2, Department of Humanities and Social Sciences, Anna University, Chennai, Orient Longmans Publication, 2008.
- Balasubramanyam, M and Anbalagan, G., Perform in English, Anuradha Publications, Kumbakonam, 2010.
- Meenakshi Raman and Sangeetha Sharma, Technical Communication: Principles and Practice, Oxford University Press, New Delhi, 2004.
- KiranmaiDutt, P.et al., A Course on Communication Skills, Edition Foundation Books, New Delhi, 2007.
- Ashraf Rizvi, M., Effective Technical Communication, Tata McGraw Hill Publication, New Delhi, 2008.
- Geoffrey Leech, Jan Swartvik, 'A Communicative Grammar of English', ELBS Longman.
- Norman and Lewis, 'English Made Easy', Oxford Publication.
- E- Writing: 21st –Century Tools for Effective Communication, Dianna Booher, Macmillan India Ltd., 2007, ISBN – 1403-93202-6



ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

Mid Semester Exam(MSE) Attendance Simulation Tasks	40 marks 05 (80%) 15 Marks	
End Semester Exam(ESE)	40 marks	

SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% in internal component or less than 40% in the end semester will be considered for supplementary assessment in the respective components (i.e. internal component or end semester) of semester concerned. Students must make themselves available during the supplementary examination period to take up the respective components (internal component or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.

Practical Work Report/Laboratory Report:

A report on the practical work is due the subsequent week after completion of the class by each group.

Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of 2% of the maximum mark per calendar day

Format

All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.

Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.

Plagiarism - Plagiarism is not acceptable and may result in the imposition of severe penalties. Plagiarism is the use of another person's work, or idea, as if it is his or her own



- if you have any doubts at all on what constitutes plagiarism, please consult your Course coordinator or lecturer. Plagiarism will be penalized severely.

Do not copy the work of other students.

Do not share your work with other students (except where required for a group activity or assessment.

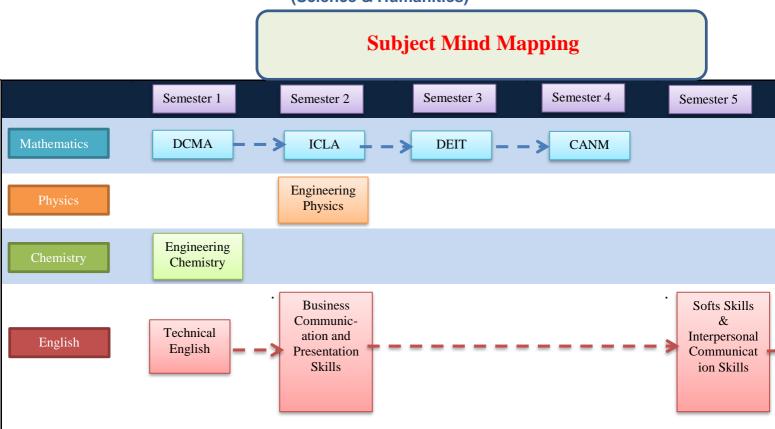
Course schedule(subject to change)

(Mention quiz, assignment submission, breaks etc as well in the table under the Teaching Learning Activity Column)

Week #	Topic & contents	CO Addressed	Teaching Learning Activity (TLA)
Weeks 1	Ice Breaking Activity	1	Written test and ABL
Weeks 2	Non Verbal Communication	1	Discussion and ABL
Week 3	Role Play	2	ABL
Week 4	Phonetics	2	Chalk & Talk
Week 5	Vocabulary	2	ABL
Week 6	Listening Theory and practice	3	ABL
Week 7	Brushing up grammar through poem	4	Discussion
Week 8	Movie Viewing& Discussing book adaptation into movies and series	4	Discussion
Week 9	Writing dialogues	1	Writing
Week 10	Reading Newspaper and social media	4	Discussion
Week 11	Reading short stories & Play enactment	4	Discussion
Week 12	Reading Test	4	Writing

्रानेन प्रकाशते जगत् INDUS UNIVERSITY

PROGRAM MAP for Bachelor of Engineering (Science & Humanities)

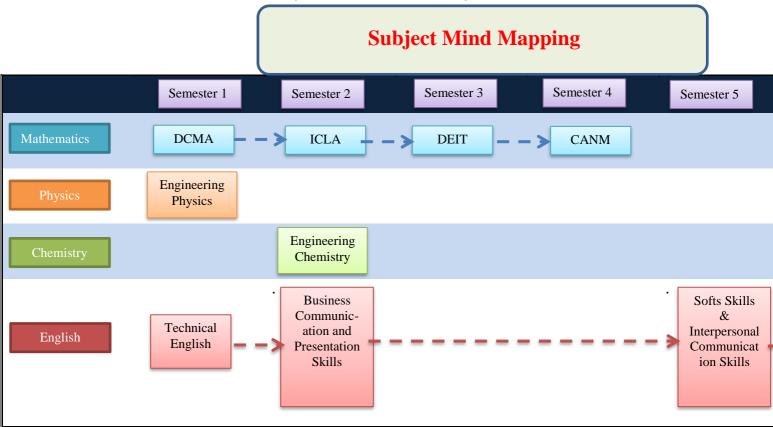


To be used for the following Branches: Information & Technology, Computer, Computer Science, Electrical, Electrics & Communication

ागेन प्रकाशते जगत् INDUS UNIVERSITY

PROGRAM MAP for Bachelor of Engineering

(Science & Humanities)



To be used for the following Branches: Automobile; Metallurgy; Civil; Mechanical



Teaching Scheme: Technical Communication

	Subject: Technical Communication							
Program: B.Tech. All Branches				Subject Code: EN0111		Semester: I	Semester: I	
								-
	Teaching	Scheme		Ex	amination Eva	luation Schen	ne	
Lecture		Practical	Credits	University Theory Examination	University Practical Examination	Continuous Internal Evaluation (CIE)- Theory	Continuous Internal Evaluation (CIE)- Practical	Total
1	2	0	3	40	0	60	0	100

Course Objectives:

- To help students develop comprehension and soft skills.
- To increase student's ability to improve and utilize the technical skills necessary for reading and writing.
- To improve students' communication skills in both technical and professional contexts.

Course Content:

Unit 1: Listening

- Icebreaking Switch Introduction
- Icebreaking Past, Present and Future
- Listening/ Cloze Test 1
- Listening/ Ted Talk
- Listening/ Josh Talk
- Listening/Celebrity Interviews
- Listening/ News Hour debates

Unit 2: Speaking

- Introduction and Polite Conversations
- Situational Dialogues
- Role Play
- Body Language
- Group Discussion
- JAM Sessions



Unit 3: Language Focus

- Teaching Remedial Grammar through Poem
- Vocabulary Building/Describing Words
- Movie Viewing and Discussion
- Book and Movie Adaptation
- Match Commentary and Review

Unit 4: Reading Skills

- Reading/Reading with Vocalic
- Read and Tweet
- Skimming/Scanning the Newspapers
- Reading/Play Enactment
- Reading/Short Stories

Reference Books:

- English for Engineers and Technologists, Volumes 1 and 2, Department of Humanities and Social Sciences, Anna University, Chennai, Orient Longmans Publication, 2008.
- Balasubramanyam, M and Anbalagan, G., Perform in English, Anuradha Publications, Kumbakonam, 2010.
- Meenakshi Raman and Sangeetha Sharma, Technical Communication: Principles and Practice, Oxford University Press, New Delhi, 2004.
- KiranmaiDutt, P. et al., A Course on Communication Skills, Edition Foundation Books, New Delhi, 2007.
- Ashraf Rizvi, M., Effective Technical Communication, Tata McGraw Hill Publication, New Delhi, 2008.
- Geoffrey Leech, Jan Swartvik, 'A Communicative Grammar of English', ELBS Longman.
- Norman and Lewis, 'English Made Easy', Oxford Publication.
- E- Writing: 21st –Century Tools for Effective Communication, Dianna Booher, Macmillan India Ltd., 2007, ISBN – 1403-93202-6
- R. K Bansal, spoken English for India (Orient Longman, Madras, 1972.

Web resources/ MOOCs:



Grammar Clauses: <u>https://www.khanacademy.org/humanities/grammar/syntax-sentences-and-clauses</u>

Parts of Speech Conjunctions & Prepositions:

https://www.khanacademy.org/humanities/grammar/parts-of-speech-the-preposition-and-the-conjunction

Nouns: https://www.khanacademy.org/humanities/grammar/parts-of-speech-the-noun

Verbs: https://www.khanacademy.org/humanities/grammar/parts-of-speech-the-verb

Pronouns: <u>https://www.khanacademy.org/humanities/grammar/parts-of-speech-the-pronoun</u>

Adjectives & Adverbs: <u>https://www.khanacademy.org/humanities/grammar/parts-of-speech-the-modifier</u>

Syntax: Conventions of Standard English:

https://www.khanacademy.org/humanities/grammar/syntax-conventions-of-standardenglish



Name of Institute: Indus Institute of Sciences, Humanities and Liberal Studies (IISHLS) Name of Faculty: Dr. Lokesh Mohan

Course code: PH0011

Course name: Engineering Physics

Pre-requisites: 12th Std Physics (Calculus, Vector analysis, Introduction of Electromagnetism, Quantum concept) Credit points: 04 Offered Semester: I

Course Coordinator (weeks 01 - 17)

Full Name: Dr. Manisha Vithalpura Department with sitting location: Physics Department, Physics lab Telephone: 3314 (sitting location), 7874636405 (Mobile) Email: manishavithalpura.gd@indusuni.ac.in Consultation times: 1:30 pm to 4:00 pm (Tuesday)

Course Lecturer (weeks 01 - 17)

Full name: Dr. Lokesh Mohan Department with sitting location: Physics Department, staff room, fourth floor, Bhanwar building Telephone: 6360912375 (Mobile) Email: lokeshmohan.ishls@indusuni.ac.in Consultation times: 1:30 pm to 4:45 pm (Friday)

Students will be contacted throughout the Session via Mail with important information relating to this Course.

Course Objectives

By participating in and understanding all facets of this Course a student will:

- To describe the basic laws of Physics, mathematical foundations and Engineering theory and to apply the knowledge in modeling and designing a real-world problem (fundamental engineering analysis skills).
- 2) To analyze a problem, identify and formulate using the concept of physics and to solve engineering problem (engineering problem solving skills).
- 3) To analyze and interpret experimental data using concepts of Physics (information retrieval skills).



 To analyze and use current techniques, skills and tools necessary for Physics and engineering practice (practical engineering analysis skills).

Course Outcomes (CO)

- 1. To understand the basic concept of physics in the engineering field
- 2. To analyze a problem, identify and formulate using the concept of physics and to solve engineering problem
- 3. To understand the properties of dielectric and magnetic material and their applications in electric and magnetic devices
- 4. To understand the basic principle of superconductivity and ultrasound with specific applications in engineering
- 5. To analyze the concept of quantum mechanics and semiconductor physics and its applications in engineering field
- 6. To understand the optical phenomena of light like Interference and Diffraction and its application in optical devices

Course Outline

UNIT-I : Introduction to Electromagnetic

Module:1 Electrostatics & Dielectrics

Coulomb's law for distribution of charges, Gauss's law and applications, Electric field intensity, Electric flux, Electric dipole moment, Electric field due to dipole, Introduction to dielectrics, Polarizability, Types of polarization – electronic, ionic, orientational, Polarization of dielectrics, Gauss's law in presence of dielectric, Dielectric constant, Electric susceptibility and Permittivity, Internal (Local) field in dielectric, Clausius Mossotti equation (with derivation)

Module:2 Magnetism

Magnetic field, Steady current, Ampere's law, Biot-Savart law and its applications, Faradays law of Induction, Lenz's Law; Effect of magnetic field on current carrying conductor, Lorentz force.

Basic important terms and units in Magnetism, Concept and origin of magnetic moment, magnetic susceptibility, Total angular momentum, Diamagnetism, Paramagnetism, Ferromagnetism, Ferrimagnetism, Antiferromagnetism, Domain theory of Ferromagnetism, Curie temperature and hysteresis loss

UNIT-II : Superconductivity and Sound

Module 1: Superconductivity

Superconductivity: Zero resistance, Critical temperature, Meissner effect, Critical field, General properties of superconductors, Type-I and Type-II superconductors, BCS theory of Superconductor, High temperature superconductors, Applications of Superconductors: SQUID, Maglev etc.

Module 2: Sound

PH0011, Semester: I (2021)



Introduction to sound waves, Characteristics and Properties of Sound, Absorption co-efficient, Reverberation time, Sabine's formula (without derivation), Factors affecting architectural acoustics,

Introduction of Ultrasonic waves, Generation of ultrasonic waves, Detection of ultrasonic waves, Applications of Ultrasonic waves: NDT, SONAR & others.

UNIT-III : Introduction to Quantum and Semiconductor Physics

Module 1: Quantum Mechanics

Black body radiation: Planck's law; Wave nature of Particles: De-Broglie theorem, Uncertainty principle; Schrodinger's wave equation – Time independent and time dependent equations; Born interpretation, probability current; Solution of stationary-state Schrodinger equation for one dimensional problems– particle in a box

Module 2 : Introduction to solids and Semiconductor Physics

Kronig-Penny model (to introduce origin of band gap), Energy bands in solids, E-k diagram; Types of electronic materials: metals, semiconductors, and insulators, Density of states, Occupation probability, Fermi level, Effective mass. Intrinsic and extrinsic semiconductors, Dependence of Fermi level on carrierconcentration and temperature (equilibrium carrier statistics), Carrier generation and recombination, Carrier transport: diffusion and drift, p-n junction diode.

UNIT-IV : Wave Optics& Laser

Module 1: Wave optics

Huygens' principle, superposition of waves and interference of light by wave front splitting and amplitude splitting; Young's double slit experiment, Newton's rings, Farunhofer diffraction from a single slit, the Rayleigh criterion for limit of resolution and its application to vision; Diffraction gratings and their resolving power, Michelson interferometer

Module 2: Laser

Einstein's theory of matter radiation interaction and A and B coefficients; Amplification of light by population inversion, Properties of laser beams: monochromaticity, coherence, directionality and brightness; Different types of lasers: gas lasers (He-Ne), solid-state lasers (Neodymium); Applications of lasers in science, engineering and medicine.

Method of delivery

(Face to face lectures, Power Point Presentation, Self assessment, Active Learning Techniques)

Study time

(3 hours per week for lectures, 2 hours per week for Practical)



Graduate Qualities and Capabilities covered (Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department of
	Graduate Capabilities
Informed Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area of study has developed and how it relates to other areas. Independent learners Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate information, using a variety of	 1 Professional knowledge, grounding & awareness Basic concept of basic physics Also application of the physics principle in engineering field 2 Information literacy, gathering & processing Critical and logical thinking is developed through numerical practice. Used various sources of the material
sources and technologies. Acknowledge the work and ideas of others.	and technology to perform the experimental part.
Problem solvers	4 Problem solving skills
Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	By practicing numerical, logical and critical thinking will be developed.
Effective communicators	5 Written communication
Articulate ideas and convey them effectively using a range of media. Work collaboratively and engage	Conducting frequent unit test will develop their written communication skill
with people in different settings. Recognize how culture can shape communication.	6 Oral communication Arranging presentation on different physics topics throughout the semester
	7 Teamwork Group discussion in class and lab is arranged
Responsible Understand how decisions can affect others and make ethically informed choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	10 Sustainability, societal & environmental impact



Practical work:

Experiment no.	Title				
1	Dielectric constant: To determine the dielectric constant				
	of a dielectric substance.				
2	To determine the magnetic field at the center of a coil				
	and its variation with distance and radius of the coil.				
3	To verify the Faraday's law of electromagnetic				
	induction.				
4	Hysteresis loss: To determine the Hysteresis loss in a				
	Ferromagnetic material.				
5	Ultrasonic Interferometer: To determine the wavelength				
	and velocity of ultrasonic wave through ultrasonic				
	interferometer.				
6	Planck's Constant: To determine the Planck's Constant				
	using LED				
7	To study the V-I characteristics of p-n junction diode				
8	To verify the Inverse Square Law using Photocell				
9	To determine the refractive index of prism using				
	Spectrometer				
10	Resolving power of grating: To determine resolving				
	power of a diffraction grating.				
11	Newton's Ring: To determine the wavelength of				
	monochromatic light				
12.	Determination of Wavelength of Laser: To determine				
	the wavelength of LASER using diffraction grating.				

Lecture/tutorial times

Example:	Lecture : Monday time: 3.20 pm to 4.10 pm Thursday time: 9.00 am to 9.55 am Friday time: 11.00 am to 11.55 am
	Practicals : Wednesday: batch-1 time: 1.30 pm to 3.20 pm Thursday : batch-2 time: 11.00 am to 12.50 pm



The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the Course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for mid and end semester examinations.

Details of referencing system to be used in written work

Unit test will be conducted in the classes and test papers will be kept with course coordinator for the future reference.

Text books

- 1. Engineering Physics by H K Malik, A K Singh, Tata Mc Graw-Hill Education Pvt. Ltd., 2nd edition, 2018, ISBN: 978-93-5260-695-5
- 2. Engineering Physics by D.K. Bhattacharya, Poonam Tandon,Oxford University Press, first published, 2015, ISBN-13:978-0-19-945281-1

Additional Materials

- Engineering Physics; Fundamentals and Modern applications by P. Khare& A. Swarup, Jones & Bartlett Learning, 2009, ISBN-13: 978-0763773748
- 2. A textbook of Engineering Physics by S.O. Pillai and Sivakami,New Age International, Third edition, 2011, ISBN:978-81-224-3162-9
- 3. An introduction to Electrodynamics by David Griffiths, Pearson Education, 3th edition, 1999, ISBN: 9780138053260
- 4. Optics by A. Ghatak, McGraw-Hill Education India Private Limited, 6th edition, 2017, ISBN-13:978-9339220907
- 5. Engineering Electromagnetics by W H Hayt& J A Buck, McGraw-Hill Education, 8th edition, 2017, ISBN-13:978-9339203276
- 6. Engineering Physics by K. Rajagopal, Prentice Hall of India Pvt. Ltd., 2007, ISBN: 9788120332867
- A Textbook of Engineering Physics by M. N. Avadhanulu, P. G. Khirsagar, S.Chand Pub., Revised edition, 1992, ISBN: 9788121908177
- University Physics, Sears and Zemansky, Pearson Education India, 13th edition, 2013, ISBN-13:978-8131790274

Web resources:



- 1. **Topics: Acoustics & Optics**: <u>http://www.nptel.iitm.ac.in/courses/Webcourse-</u> <u>contents/IIT%20Guwahati/engg_physics/index_cont.htm</u>
- 2. Course: Engineering Physics: http://www.nptelvideos.in/search?g=engineering+physics
- 3. **Topic: Laser:**http://science.howstuffworks.com/laser1.htm
- 4. Topic: Optics: http://www.pitt.edu/~poole/physics.html#light
- 5. Topic: Magnetism: <u>https://www.khanacademy.org/science/physics/magnetic-forces-and-magnetic-fields</u>
- 6. Topic: Interference: https://www.khanacademy.org/science/physics/light-waves
- 7. Topic: Quantum Mechanics: <u>https://ocw.mit.edu/courses/physics/8-04-quantum-physics-i-spring-2016/index.htm</u>

MOOCs: https://www.edx.org/course/subject/physics

ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

Continuous Internal Evaluation (Theory)				
Mid-Sem exam	40% , Unit-1/2, Objective (1,2,5)			
Assignment	10% , Objective (1,2,5)			
Project/Presentation	5%, Objective (1,2,6)			
Attendance 5% (end of the semester)				
Total 60% (CIE theory)				
Final exam (closed book)	40% Objectives (1-6)			

SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% in mid semester or end semester will be considered for supplementary assessment in the respective components (i.e mid semester or end semester) of semester concerned. Students must make themselves available during the supplementary examination period to take up the respective components (mid semester or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.



Practical Work Report/Laboratory Report:

Student has to complete the experiment in their respective lab in a week, will be evaluated weekly basis. A report on practical work is to be submitted after completion of the lab by each group.

Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of -% of the maximum mark per calendar day

Format

All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.

Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.

Plagiarism - Plagiarism is not acceptable and may result in the imposition of severe penalties. Plagiarism is the use of another person's work, or idea, as if it is his or her own - if you have any doubts at all on what constitutes plagiarism, please consult your Course coordinator or lecturer. Plagiarism will be penalized severely.

Do not copy the work of other students. Do not share your work with other students (except where required for a group activity or assessment.

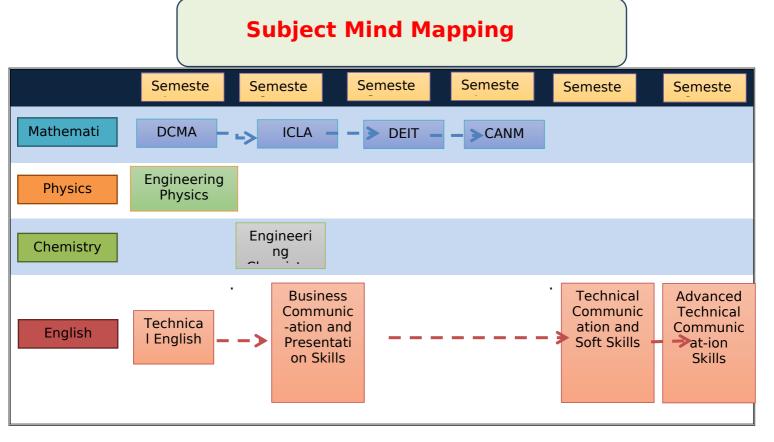
Co	ourse sched	ule(subject to change)		
	Week #	Topic & contents	CO Addressed	Teaching Learning Activity (TLA)

			ज्ञानेन प्रकाशते जगत् INDUS UNIVERSITY
Weeks 1	M-1 : Coulomb's law for distribution of charges, Gauss's law and applications, Electric current and Equation of continuity, Electric field intensity, Electric flux, Electric dipole moment, Divergence and curl of electrostatic field, Introduction to dielectrics, Polarizability, Types of polarization – electronic, ionic, orientational, Polarization of dielectrics, Gauss's law in presence of dielectric, Dielectric constant, Electric susceptibility and Permittivity,	1,2,5	PPt, chalk and board
Weeks 2	Internal (Local) field in dielectric, Clausius Mossotti equation (with derivation), M-2: Magnetic field, Steady current, Biot-Savart law, Ampere's law, Divergence and curl of static magnetic field, vector potential and calculating it for a given magnetic field using Stokes' theorem,	1,2,3	Chalk and Board
Week 3	Effect of magnetic field on current carrying conductor, Lorentz force, Faraday's law in terms of EMF produced by changing magnetic flux; Lenz's law; M-3: Basic important terms and units in Magnetism, Concept and origin of magnetic moment, magnetic susceptibility, Total angular momentum	1,2,3	Chalk and Board
Week 4	Diamagnetism, Paramagnetism, Ferromangnetism, Domain theory of Ferromagnetism, hysteresis loss, numerical,	1,2,3	Chalk and Board
Week 5	Introduction to Superconductors and its properties, types of it	1,2,4	PPT and chalk- board
		1,2,4	
Week 6	Numerical practice, Test	1,2,4	Chalk and board

			ज्ञानेन प्रकाशते जगत् INDUS UNIVERSITY
Week 7	Types-I , II superconductor, BCS theory and introduction to Sound wave	1,2,4	Ppt and chalk and board
Week 8	Audible sound and its characteristics, factor affecting and ultrasound	1,2,4	Ppt and chalk- board
Week 9	Introduction to subject, M-1 : Introduction, Black body radiation and laws if Black body radiation, De-Broglie theorem, Uncertainty principle; Schrodinger's wave equation	1,2,5	PPt, chalk and board
Week 10	Born interpretation, probability current; Solution of stationary- state, Schrodinger equation for one dimensional problems- particle in a box, Numerical practice	1,2,5	PPt, chalk and board
Week 11	Types of electronic materials: metals, semiconductors, and insulators, Density of states, Occupation probability, Fermi level, Effective mass., Intrinsic and extrinsic semiconductors, Dependence of Fermi level on carrier-concentration and temperature (equilibrium carrier statistics),	1,2,5	Chalk and board
Week 12	Carrier generation and recombination, Carrier transport: diffusion and drift, p- n junction diode & Numericals, Unit test-1	1,2,5	Chalk and board
Week 13	M-1: Wave front, Huygens' principle, superposition of waves and interference of light by wave front splitting and amplitude splitting, Young's double slit experiment, Newton's rings, Numerical, test II		PPTs and Chalk and board
Week 14	Farunhofer diffraction from a single slit, the Rayleigh criterion for limit of resolution and its application to vision;, Diffraction gratings and their resolving power, Michelson interferometer, numerical	1,2,6	PPTs and Chalk and Board

Week 15	M-2: Einstein's theory of matter radiation interaction and A and B coefficients; Amplification of light by population inversion, Properties of laser beams: mono-chromaticity, coherence, directionality and brightness; Different types of lasers: gas lasers (He-Ne), solid-state lasers (Neodymium);	1,2,6	ज्ञानेन प्रकाशते जगत् INDUS UNIVERSITY PPTs
Week 16	Applications of lasers in science, engineering and medicine., Numerical problems	1,2,6	PPTs
Week 17	Revision	1-6	РРТ

PROGRAM MAP for Bachelor of Engineering (Institute of Sciences, Humanities and Liberal Studies)



To be used for the following Branches: Automobile Engineering



Name of Institute: Indus Institute of Technology & Engineering Name of Faculty: Prof. Jainisha Patel/ Prof. Pradnya Tikhe

Course code: CV0004 Course name: Environmental Science Pre-requisites: Basic knowledge on Environmental issues. Knowledge on chemistry and laboratory equipment. Credit points: 2

Offered Semester: 1st (Information & Technology)

Course Coordinator

Full name: Prof. Pradnya Tikhe Department with siting location: 3rd Floor staff Room, Bhanwar Builing Telephone: 3006 Email: pradnyatikhe.cvl@indusuni.ac.in Consultation times: 9.00 to 5.00

Course Lecturer

Full Name: Prof. Jainisha Patel

Department with siting location: Environmental Engineering Lab, Ground floor, Extension building

Telephone: 3006

CV0004, Semester: 02 (2020)



Email: jainishapatel.cvl@indusuni.ac.in

Consultation times: 9.00 to 5.00

Students will be contacted throughout the Session via Mail with important information relating to this Course.

Course Objectives

By participating in and understanding all facets of this Course a student will:

- Understand key concepts from economic, political, and social analysis as they pertain to the design and evaluation of environmental policies and institutions.
- Appreciate concepts and methods from ecological and physical sciences and their application in environmental problem solving.
- Appreciate the ethical, cross-cultural, and historical context of environmental issues and the links between human and natural systems.
- Reflect critically about their roles and identities as citizens, consumers and environmental actors in a complex, interconnected world.

Course Outcomes (CO)

- Master core concepts and methods from ecological and physical sciences and their application in environmental problem solving.
- Master core concepts and methods from economic, political, and social analysis as they pertain to the design and evaluation of environmental policies and institutions.
- Appreciate the ethical, cross-cultural, and historical context of environmental issues and the links between human and natural systems.
- Understand the transnational character of environmental problems and ways of addressing them, including interactions across local to global scales.
- Apply systems concepts and methodologies to analyze and understand interactions between social and environmental processes.
- Reflect critically about their roles and identities as citizens, consumers and environmental actors in a complex, interconnected world.

Course Outline

To introduce students to the integrated science, engineering, design and management concepts of engineered environmental systems. The course will cover environmental regulations and



standards, environmental parameters, mass balance and natural systems, water quality management, water and wastewater treatment, air pollution control, noise pollution, and solid and hazardous waste management. Background material and laboratories in the environmental sciences and management areas will be covered. Group term papers and presentations will be required.

Method of delivery

60% of Lectures consist of Face to face lectures which are conducted in classrooms, 30% of lectures comprise of PowerPoint Presentation through which various videos and images of related topics are shown to the students, and 10% of lectures consist of hands on session.

Study time

2 hours of lectures (theory) per week.

CO-PO Mapping (PO: Program Outcomes)

	P01	PO2	PO3	PO4	PO5	PO6	P07	PO8	PO9	PO10	PO11	PO12
CO1	3	-	-	-	-	2	-	-	-	1	-	-
CO2	-	3	-	-	-	-	3	2	-	-	-	1
CO3	-	-	-	-	-	3	2	-	-	-	1	-
CO4	-	3	2	-	-	-	1	-	-	1	-	-
CO5	-	2	-	3	-	-	-	-	-	-	-	-
CO6	3	-	2	-	1	1	-	-	-	-	-	-

1-Lightly Mapped

2- Moderately Mapped 3- Highly Mapped



Blooms Taxonomy and Knowledge retention (For reference)

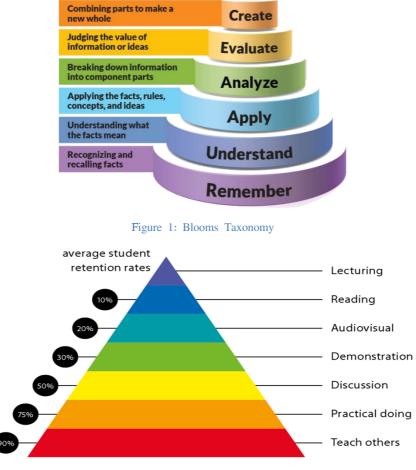


Figure 2: Knowledge retention

Graduate Qualities and Capabilities covered

(Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department ofGraduate Capabilities
Informed	1 Professional knowledge, grounding &
Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area of study has developed and how it relates to other areas.	awareness



Independent learners	2 Information literacy, gathering & processing
Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate information, using a variety of sources and technologies. Acknowledge the work and ideas of others.	
Problem solvers Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	4 Problem solving skills
Effective communicators	5 Written communication
Articulate ideas and convey them effectively using a range of media. Work collaboratively and engage with people in different settings. Recognize how culture can shape communication.	6 Oral communication7 Teamwork
Responsible Understand how decisions can affect others and make ethically informed choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	10 Sustainability, societal & environmental impact

Practical work:

Lecture/laboratory times

Lecture (IT-I)

Tuesday 09:55 to 10:50

Wednesday 11:55 to 12:50

Lecture (IT-II)



Attendance Requirements

The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the Course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for mid and end semester examinations.

Details of referencing system to be used in written work

Text books

1. Textbook of Environmental Studies for Undergraduate Courses by Erach Bharucha Second edition, 2013 Publisher: Universities Press (India) Private Ltd, Hyderabad

Reference Books:

- Basics of Environmental Studies by Prof Dr N S Varandani ,2013 Publisher: LAP Lambert Academic Publishing , Germany
- Environmental Studies by Anindita Basak ,2009 Publisher: Drling Kindersley(India)Pvt. Ltd
- Pearson Textbook of Environmental Studies by Deeksha Dave & S S Kateva, Cengage Publishers.
- Environmental Sciences by Daniel B Botkin & Edward A Keller Publisher: John Wiley & Sons.
- Environmental Studies by R. Rajagopalan, Oxford University Press
- Environmental Studies by Benny Joseph, TMH publishers
- Environmental Studies by Dr. Suresh K Dhameja, 2007 Published by: S K Kataria & Sons New Delhi
- Basics of Environmental Studies by U K Khare, 2011 Published by Tata McGraw Hill



Additional Materials

ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

Example:						
Class test	05					
Power Point Presentation	15					
Final exam (closed book)	40					

SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% in mid semester or end semester will be considered for supplementary assessment in the respective components (i.e mid semester or end semester) of semester concerned. Students must make them available during the supplementary examination period to take up the respective components (mid semester or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.

Practical Work Report/Laboratory Report:

A report on the practical work is due the subsequent week after completion of the class by each group.

Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of -% of the maximum mark per calendar day



Format

All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.

Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.

Plagiarism - Plagiarism is not acceptable and may result in the imposition of severe penalties. Plagiarism is the use of another person's work, or idea, as if it is his or her own - if you have any doubts at all on what constitutes plagiarism, please consult your Course coordinator or lecturer. Plagiarism will be penalized severely.

Do not copy the work of other students.

Do not share your work with other students (except where required for a group activity or assessment)



Course schedule (subject to change)

.

Week #	Topic & contents	CO Addressed	Teaching Learning Activity (TLA)
Weeks 1	Environment: Levels of organizations in environment, Structure and functions in an ecosystem Biosphere: Its Origin and distribution on land, in water and in air, Ecosystem and functioning of Ecosystem	1,3	PPT and BB
Weeks 2	Natural Resources	2,1,4	PPT
Week 3	Biodiversity and its conservation	3,2	BB and PPT
Week 4	Environmental Pollution:	2,4,6	BB
Week 5	Environmental Pollution, and Environmental Prevention	1,3	BB and PPT
Week 6	Concept of clean Environment, man-made disasters, Case Studies of Indian Studies, Urban flooding, Heat Island Effect	2,4	BB
Week 7	Management of pollution Technologies for Pollution Control Environmental Monitoring	1,5	BB and PPT
Week 8	Research and Policies relating to environmental Protection	2,4,6	BB and PPT
Week 9	Concept of sustainability and environmental Sustainability Index Environmental Ethics, people's participation	2,4,6	BB and PPT
Week 10	Green Business (Profitability in managing Environment)	1,5,6,3	BB and PPT



	Sustainable Development,		
Week 11	Public awareness, public participation	5.6.4	BB and PPT
Week 12	Revision		РРТ



Subject: Environmental Science										
Program	n: B.Tech	. offered	by Civil I	Dept.	ept. Subject Code:CV0004 Semeste			r: I/II		
Teaching Scheme Examination Evaluation Scheme										
Lecture	Tutoria l	Practica 1	Credits	Continuous Internal Evaluation (CIE)- Theory	Continuous Internal Evaluation (CIE)-Practi cal	University Theory Examinatio n	University Practical Examination	Total		
02	00	-	02	24/60	-	16/40	-	100		

Course Objectives:

- Understand key concepts from economic, political, and social analysis as they pertain to the design and evaluation of environmental policies and institutions.
- Appreciate concepts and methods from ecological and physical sciences and their application in environmental problem solving.
- Appreciate the ethical, cross-cultural, and historical context of environmental issues and the links between human and natural systems.
- Reflect critically about their roles and identities as citizens, consumers and environmental actors in a complex, interconnected world.

Course Outcomes:

- Master core concepts and methods from ecological and physical sciences and their application in environmental problem solving.
- Master core concepts and methods from economic, political, and social analysis as they pertain to the design and evaluation of environmental policies and institutions.
- Appreciate the ethical, cross-cultural, and historical context of environmental issues and the links between human and natural systems.
- Understand the transnational character of environmental problems and ways of addressing them, including interactions across local to global scales.
- Apply systems concepts and methodologies to analyze and understand interactions between social and environmental processes.
- Reflect critically about their roles and identities as citizens, consumers and environmental actors in a complex, interconnected world.



COURSE CONTENTS

<u>Unit I</u>

Concepts of Environmental Sciences: Environment: Levels of organizations in environment, Structure and functions in an ecosystem;

Biosphere: Its Origin and distribution on land, in water and in air, Ecosystem and functioning of Ecosystem.

Natural Resources: Renewable and Non-renewable Resources, Forests, water, minerals, Food and land (with example of one case study); Energy, Growing energy needs, energy sources (conventional and alternatives), State of Indian Environment.

Biodiversity and its conservation:

Biodiversity at global, national and local levels; India as a mega-diversity nation; Threats to biodiversity (biotic, abiotic stresses), and strategies for conservation.

Unit II

Environmental Pollution:

Types of pollution- Air, water (including urban, rural, marine), soil, noise, thermal, nuclear, Solid Waste (Organic and Inorganic), Concept of clean Environment; Case Studies of Indian Studies.

Introduction to mam-made disasters(Urban Flooding, Heatwave)

Unit III

Pollution prevention:

Management of pollution- Rural/Urban/Industrial waste management [with case study of any one type, e.g., power (thermal/nuclear), fertilizer, tannin, leather, chemical, sugar], Solid/Liquid waste management, disaster management.

Environmental Monitoring:

Monitoring- Identification of environmental problem, tools for monitoring (remote sensing, GIS); Sampling strategies- Air, water, soil sampling techniques

Technologies for Pollution Control: Bio-remediation, phytoremediation, bio-pesticides, bio-fertilizers;

Legal issues: Environmental legislation (Acts and issues involved), Environmental Protection Acts : (Swacch Bharat Abhiyan, state Action Plans)

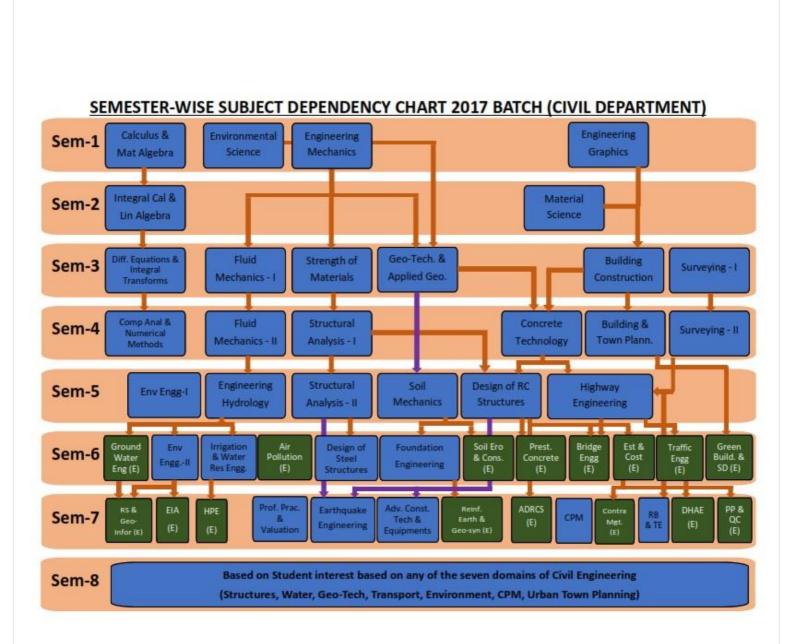
Research and Policies relating to environmental Protection

Unit IV



Social Issues and the Environment:

Concept of sustainability and Sustainable Development, environmental Sustainability Index, Environmental Ethics, Public awareness and people's participation, Green Business (Profitability in managing Environment)



Page 13 of 13



Name of Institute: IITE Name of Faculty: Mr.Ajay Sikdar

Course code: MME0103

Course name: Material Science Pre-requisites: Physics, Chemistry Credit points: 3 Offered Semester: 1st sem

Course Coordinator (weeks XX - XX)

Full Name: Mr.Ajay Sikdar Department with sitting location: Third Floor staff room Telephone:99798 53458 Email: ajaysikdar.mt@indusuni.ac.in Consultation times: Monday and Tuesday 4:15 – 5:00PM

Course Lecturer (weeks xx - XX)

Full Name: Mr.Ajay Sikdar Department with sitting location: Third Floor staff room Telephone:99798 53458 Email: ajaysikdar.mt@indusuni.ac.in Consultation times: Monday and Tuesday 4:15 – 5:00PM

Students will be contacted throughout the Session via Mail with important information relating to this Course.

Course Objectives

By participating in and understanding all facets of this Course a student will:

1. To evaluate different materials for engineering applications.

2. To categorize material according to their properties and requirement.

3. To classify materials and understand the importance of each material in order to find applications in other fields of engineering.

Course Outcomes (CO)

CO1- Students will be able to define material Science. [BT-11]

CO2- Students will be able to classify various Engineering Materials. [BT-2]

CO3- Students will be able to identify various Engineering Materials.[BT-3]

CO4- Students will be able to compare various Engineering Materials and their properties. [BT-4]

CO5-Students will be able to recommend material for application. [BT-5]

CO6- Students will be able discuss the criteria for selection of Materials.[BT-6]



Course Outline

The proposed course deals with various surface preparation and coating methods. .

Method of delivery

(Face to face lectures, self study material, Active Learning Techniques)

Study time

3 lectures

	Introduction to Engineering Materials (MME0101)											
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	2	2	2	1	1	1	-	-	-	-	1	2
CO2	2	1	1	1	2	1	-	-	-	-		1
CO3	2	1	1	2	1	2	-	-	-	-	2	1
CO4	1	2	2	1	2	-	-	-	-	-	-	2
CO5	1	2	2	1	-	-	-	-	-	-	-	-
CO6	2	2	1	-	-	-	-	-	-	-	1	1
MME0101	1.66	1.66	1.5	1	1	1.33	-	-	-	-	1.33	1.4

CO-PO Mapping (PO: Program Outcomes)

1-Lightly Mapped 2- Moderately Mapped 3- Highly Mapped

Blooms Taxonomyand Knowledge retention(For reference)

(Blooms taxonomy has been given for reference)



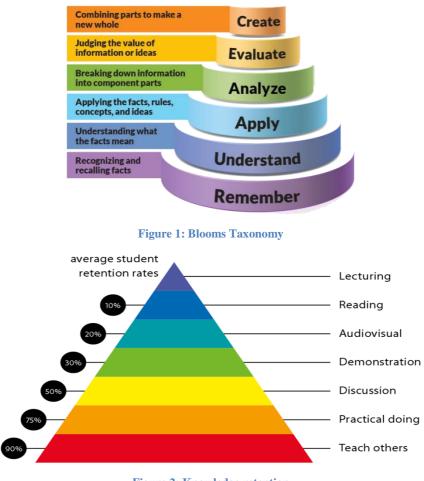


Figure 2: Knowledge retention

Graduate Qualities and Capabilities covered

(Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department ofGraduate Capabilities
Informed Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area of study has developed and how it relates to other areas.	1 Professional knowledge, grounding & awareness



Independent learners Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate information, using a variety of sources and technologies. Acknowledge the work and ideas of others.	2 Information literacy, gathering & processing
Problem solvers Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	4 Problem solving skills
Effective communicators Articulate ideas and convey them effectively using a range of media. Work collaboratively and engage with people in different settings. Recognize how culture can shape communication. Responsible Understand how decisions can affect others and make ethically informed	5 Written communication 6 Oral communication 7 Teamwork 10 Sustainability, societal & environmental impact
choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	

Practical work:

NA

Lecture/tutorial times

(Give lecture times in the format below)



Lecture	Monday	09.55 – 10.50 am
Lecture	Monday	02.30 – 03.30 pm
Lecture	Tuesday	09.00 – 09.55 am

Attendance Requirements

The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the Course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for mid and end semester examinations.

Details of referencing system to be used in written work

Text Book(s)

1. O. P. Khanna, "Material Science and Metallurgy", Dalpat Rai Publications, 2nd Edition, 2014, ISBN: 9789383182459.

2. R. K Rajput, "Engineering Materials", S. Chand Publications, 4th Edition, 2000, ISBN: 9788121919609.

3. W.D. Callister, "Material Science & Engineering – An Introduction", John Wiley Publishers, 7th Edition, 2007, ISBN: 9780471736967.

Reference Books

1. J. Shackelford, "Introduction to Materials Science for Engineers", Pearson-Prentice Hall Publications, 8th Edition, 2006, ISBN: 8131700909.

2. L.H. Vanvlack, "Elements of Materials Science and Engineering", Pearson Education India, 6th Edition, 2002, ISBN: 8131706001.

3. D. Swarup, "Elements of Metallurgy", Rastogi Publications, 2005, ISBN: 8171338135.

4. V. Raghavan, "Materials Science and Engineering – A First Course", Prentice Hall India Learning Private Limited, 6th Edition, 2015, ISBN: 8120350928.

Digital Learning Resources

www.nptel.ac.in

www.edx.org



ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

Example:					
MSE	40%				
Quiz 1	10%				
Quiz 2	10%				
Final exam (closed book)	40%				

SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% CIE or end semester will be considered for supplementary assessment in the respective components (i.e CIE or end semester) of semester concerned. Students must make themselves available during the supplementary examination period to take up the respective components (mid semester or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.

Practical Work Report/Laboratory Report: (NA)

A report on the practical work is due the subsequent week after completion of the class by each group.

Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of -1% of the maximum mark per calendar day

Format

All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.



Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.

Plagiarism - Plagiarism is not acceptable and may result in the imposition of severe penalties. Plagiarism is the use of another person's work, or idea, as if it is his or her own - if you have any doubts at all on what constitutes plagiarism, please consult your Course coordinator or lecturer. Plagiarism will be penalized severely.

Do not copy the work of other students.

Do not share your work with other students (except where required for a group activity or assessment)



Course schedule (subject to change)

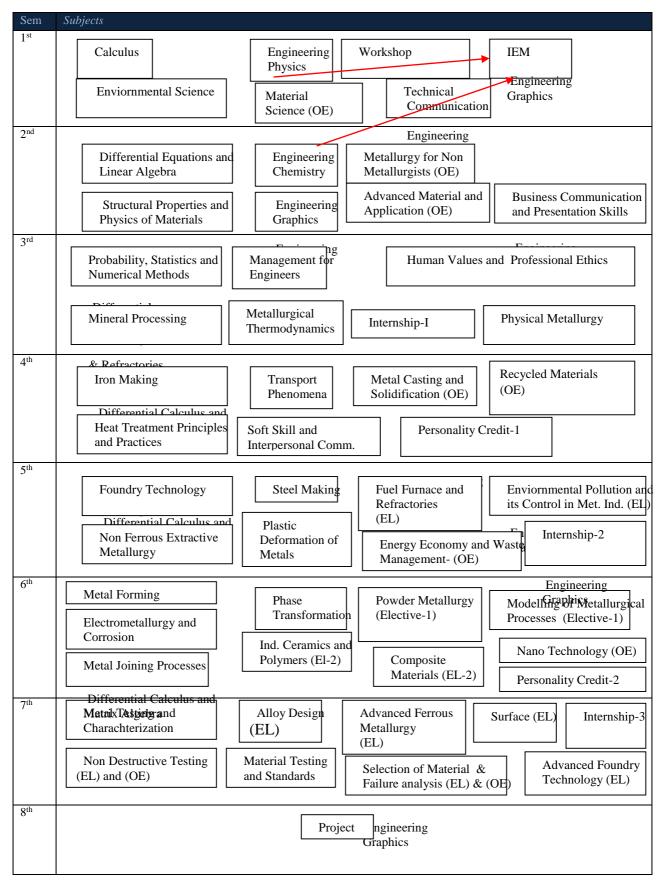
Week #	Topic & contents	CO Addressed	Teaching Learning Activity (TLA)
Weeks 1	Introduction about subject, requirement of different materials, Classification of Engineering materials		BB, PPT
Weeks 2	Properties of engineering materials, Criteria for selection of materials for engineering application.		BB, PPT
Week 3	Structure of crystalline solids; Lattices, unit cells; atom size, co-ordination number, atomic packing factor., Indexing of directions and planes, notations		BB, PPT
Week 4	Interplanar spacing and angles, Crystalstructure analysis - Bragg's law for X-ray diffraction		BB, PPT
Week 5	Pig iron, cast iron, carbon steel, alloy steels- Classification, properties, composition and applications.		BB, PPT
Week 6	Important non-ferrous metals Zn, Sn, Mg, Ti, Ni, Composition, properties, classification and applications		BB, PPT
Week 7	Zn, Sn, Mg, Ti, Ni, Composition, properties, classification and applications		BB, PPT
Week 8	Introduction, Simple crystal structure, Classification- Traditional (clay-products, refractories, abrasives, cement) and Engineering Ceramics		BB, PPT
Week 9	Glass Ceramics, Properties of ceramics, Application of Ceramics, Glasses, Glass structure, Properties and application of Glass, Types of glass		BB, PPT
Week 10	Introduction, Classification and forms of Polymers, Thermosetting & thermoplastic polymer, types of polymerizations,		BB, PPT



	Molecular weight, Plastics, Natural rubber and synthetic rubber, Applications of polymeric materials	
Week 11	Introduction, Classification & Applications, Dispersion-strengthened, Composites, Particulate Composites, Fiber-reinforced Composites: Influence of Fiber Length, Influence of Fiber Orientation and Concentration, The Fiber Phase	BB, PPT
Week 12	The Matrix Phase, Polymer-Matrix. Composites, Metal-Matrix Composites, Ceramic - Matrix Composites, Carbon– Carbon Composites, Processing of Fiber- Reinforced Composites	BB, PPT
Week 13	Smart materials (Shape memory material, Piezo electric material) Photoconductors,	BB, PPT
Week 14	Bio-materials, Nano materials, Dielectric materials	BB, PPT
Week 15	magnetic materials, metamaterials, Cryogenics, Optical Fiber	BB, PPT



Program Mapping (Metallurgy Engineering Department)



Nano Tec



	Subject: Introduction to Engineering Materials							
Program: B. Tech in Metallurgical Engineering				neering	eering Subject Code: MME0103			ter: I
Teaching Scheme (Hours per week)			Exam	Examination Evaluation Scheme (Marks)				
Lecture	Tutorial	Practical	Credits	University Theory Examination	University Practical Examination	Continuous Internal Evaluation (CIE)- Theory	Continuous Internal Evaluation (CIE)- Practical	Total
3	0	0	3	16/40	0	24/60	0	100

Course Objectives

1. To evaluate different materials for engineering applications.

2. To categorize material according to their properties and requirement.

3. To classify materials and understand the importance of each material in order to find applications in other fields of engineering.

CONTENTS

<u>UNIT-I</u>

Materials

Introduction, Engineering requirement of different materials, Classification of Engineering materials, Properties of engineering materials, Criteria for selection of materials for engineering application. **Crystal Physics**

Structure of crystalline solids; Lattices, unit cells; atom size, co-ordination number, atomic packing factor., Indexing of directions and planes, notations, Interplanar spacing and angles, Crystal structure analysis - Bragg's law for X-ray diffraction.

UNIT-II

Ferrous metals & Alloys

Pig iron, cast iron, carbon steel, alloy steels- Classification, properties, composition and applications. Non-Ferrous Metals & Alloys

Important non-ferrous metals (Al, Cu, Pb, Zn, Sn, Mg, Ti, Ni,), Non-ferrous alloys (Cu alloys, Al alloys, Mg-alloys, Ni-alloys) – Composition, properties, classification and applications.

UNIT-III

Ceramics

Introduction, Simple crystal structure, Classification- Traditional (clay-products, refractories, abrasives, cement) and Engineering Ceramics - Glass Ceramics, Properties of ceramics, Application of Ceramics, Glasses, Glass structure, Properties and application of Glass, Types of glass.

Polymers

Introduction, Classification and forms of Polymers, Thermosetting & thermoplastic polymer, types of polymerizations, Molecular weight, Plastics, Natural rubber and synthetic rubber, Applications of polymeric materials.

[10 hours]

[10 hours]

[10 hours]



[10 hours]

UNIT-IV

Composites

Introduction, Classification & Applications, Dispersion-strengthened, Composites, Particulate Composites, Fiber-reinforced Composites: Influence of Fiber Length, Influence of Fiber Orientation and Concentration, The Fiber Phase, The Matrix Phase, Polymer-Matrix.

Composites, Metal-Matrix Composites, Ceramic - Matrix Composites, Carbon–Carbon Composites, Processing of Fiber-Reinforced Composites.

Advanced Materials

Smart materials (Shape memory material, Piezo electric material) Photoconductors, Bio-materials, Nano materials, Dielectric materials, magnetic materials, metamaterials, Cryogenics, Optical Fiber.

Course Outcomes

1. To apply the fundamentals of mass, matter and materials from daily life.

2. To acquaint the student with applications and properties of materials used from engineering aspects.

3. To apply student's knowledge about advanced materials to be used in futuristic applications.

Text Books

1. O. P. Khanna, "Material Science and Metallurgy", Dalpat Rai Publications, 2nd Edition, 2014, ISBN: 9789383182459.

2. R. K Rajput, "Engineering Materials", S. Chand Publications, 4th Edition, 2000, ISBN: 9788121919609.

3. W.D. Callister, "Material Science & Engineering – An Introduction", John Wiley Publishers, 7th Edition, 2007, ISBN: 9780471736967.

Reference Books

1. J. Shackelford, "Introduction to Materials Science for Engineers", Pearson-Prentice Hall Publications, 8th Edition, 2006, ISBN: 8131700909.

2. L.H. Vanvlack, "Elements of Materials Science and Engineering", Pearson Education India, 6th Edition, 2002, ISBN: 8131706001.

3. D. Swarup, "Elements of Metallurgy", Rastogi Publications, 2005, ISBN: 8171338135.

4. V. Raghavan, "Materials Science and Engineering – A First Course", Prentice Hall India Learning Private Limited, 6th Edition, 2015, ISBN: 8120350928.

Web Resources

1. MOOC Course on "Materials Science and Engineering" (https://www.edx.org/course/materialsscience-engineering-misisx-mse1x)



Name of Institute: IITE Name of Faculty: Dr. Jigna Panchal

Course code: MA0112

Course name: Multivariable Calculus

Credit points: 4 Offered Semester: I

Course coordinator (weeks 1 - 15)

Full name: Prof.Kinjal Shah Department with siting location: Mathematics Department, ISHLS,4th floor Bhanwar building, Indus University, Ahmadabad Telephone: 3425 Email: kinjalshah.gd@indusuni.ac.in Consultation times: 10 am - 12 pm

Course lecturer (weeks 1 - 15)

Full name: Dr. Jigna Panchal Department with siting location: Mathematics, 4th floor, Bhanwar Building. Telephone: 3424 Email:jignapanchal.gd@indusuni.ac.in Consultation times: Wednesday: 2:00 to 4:00 PM

Students will be contacted throughout the session via mail with important information relating to this course.

Course Objectives

By participating in and understanding all facets of this course a student will:

- 1) To provide mathematical knowledge and skills needed to support their concurrent and subsequent engineering studies.
- 2) To provide an ability to apply knowledge of basic science and engineering fundamentals.
- 3) To provide an ability to undertake problem identification, formulation and solution.
- 4) To provide an ability to analyze different mathematical models within science and technology and work creatively, systematically and critically.
- 5) To provide an ability to find strategies for the solution of different types of mathematical models using knowledge about the possibilities and limitations of the different methods and tools.
- 6) To provide an ability to develop abstract, logical and critical thinking and the ability to reflect critically upon their work and work of others.



7) To provide an ability to insight their strengths and weakness as learners and to appreciate the value of errors or mistakes as powerful motivators to enhance learning and understanding.

Course Outcomes (CO)

Upon the successful completion of the course, students will be able to:

	Develop mathematical knowledge and skills needed to support their concurrent and subsequent engineering studies.	BT-4
C O : 2	Describe an ability to undertake problem identification, formulation and solution.	BT-1
	Evaluate the solution of different types of mathematical models using knowledge about the possibilities and limitations of the different methods and tools.	BT-5
	Apply the knowledge of basic science and engineering fundamentals.	BT-3
CO : 5	Explain the concept of partial differentiation, differential calculus and vector algebra.	BT-2
C O : 6	Create different mathematical models within science and technology and work creatively, systematically and critically.	BT-6

Course Outline

Basics of Partial Derivatives and its applications, Vector Differential & integral Calculus.

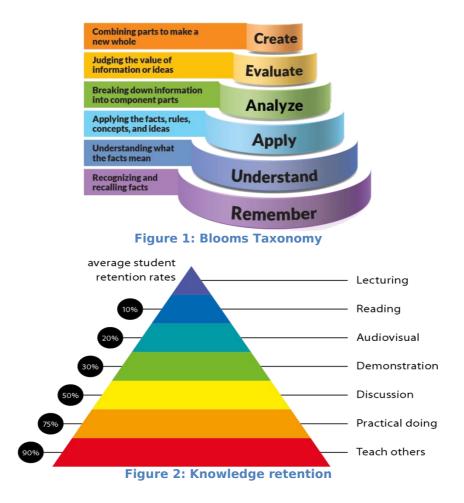
Method of delivery Chalk & Talk

Study time

4 Hrs/week

Blooms Taxonomy and Knowledge retention (For reference) (Blooms taxonomy has been given for reference)





Graduate Qualities and Capabilities covered

(Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department of Metallurgical Graduate Capabilities
Informed Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area of study has developed and how it relates to other areas.	1 Professional knowledge, grounding & awareness
Independent learners Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate	2 Information literacy, gathering & processing



information, using a variety of sources and technologies. Acknowledge the work and ideas of others.	
Problem solvers Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	4 Problem solving skills
Effective communicators Articulate ideas and convey them	5 Written communication 6 Oral communication
effectively using a range of media. Work collaboratively and engage with people in different settings. Recognize how culture can shape communication.	7 Teamwork
Responsible Understand how decisions can affect others and make ethically informed choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	10 Sustainability, societal & environmental impact

Practical work:

Not Applicable

Lecture/tutorial times

(Give lecture times in the format below)

Example:			
Lecture	Monday 2.	25 pm - 3:20 pm	Room: LH-13
Lecture	Tuesday	1:30 pm - 2:25 pm	Room: LH-13
Lecture	Thursday	9:55 am - 10:50 am	Room: LH-14
Tutorial	Wednesday	9:00 am - 9:55 am	Room: LH-14
	-		



Attendance Requirements

The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for semester examinations.

Details of referencing system to be used in written work

Text books

B. V. Ramana, "Higher Engineering Mathematics", Tata McGraw Hill.

Additional Materials

Reference Books:

- 1. Erwin Kreyszig, "Advanced Engineering Mathematics" (8th Edition), Wiley Eastern Ltd., New Delhi.
- 2. Dr. B.S. Grewal, "Higher Engineering Mathematics", Khanna Publishers, New Delhi
- 3. Murray Spiegel , "Advanced Mathematics for Engineering & Science: Schaum's Outline Series", Tata McGraw Hill Publication
- 4. MerelC Potter, J.L. Goldberg, "Advanced Engineering Mathematics" (3rd Edition), Oxford India Publication.

ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

	Internal Exam	am 60% Objective (20 Marks- Class Test-1, 20 Test-2, 5 Marks- Attenda Assignment, 10 Marks Quiz)	
Su	Final exam (closed book)	40%	Objectives (1-5)



SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% in internal component or less than 40% in the end semester will be considered for supplementary assessment in the respective components (i.e internal component or end semester) of semester concerned. Students must make themselves available during the supplementary examination period to take up the respective components (internal component or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.

Practical Work Report/Laboratory Report:

A report on the practical work is due the subsequent week after completion of the class by each group.

Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of -% of the maximum mark per calendar day

Format

All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.

Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.

Plagiarism - Plagiarism is not acceptable and may result in the imposition of severe penalties. Plagiarism is the use of another person's work, or idea, as if it is his or her own - if you have any doubts at all on what constitutes plagiarism, please consult your Course coordinator or lecturer. Plagiarism will be penalized severely.

Do not copy the work of other students. Do not share your work with other students (except where required for a group activity or assessment)



Course schedule (subject to change)

(Mention quiz, assignment submission, breaks etc as well in the table under the Teaching Learning Activity Column)

Week #	Topic & contents	CO Addressed	Teaching Learning Activity (TLA)
Weeks 1	Partial Differentiation Variable Treated as Constant	Apply the knowledge of multivariable calculus and partial differentiation for solving various practical & engineering problems.	Chalk & Talk, Assignment Submission , Unit Test.
Weeks 2	Total Derivative, Partial Differentiation of Composite Functions: Change of Variable,Jacobian	Apply the knowledge of multivariable calculus and partial differentiation for solving various practical & engineering problems.	Chalk & Talk, Assignment Submission , Unit Test.
Week 3	Maxima and Minima of Functions of two Variables: with and without constraints, Lagrange's Method of Undetermined Multipliers	Apply the basic concept of partial derivatives and their applications.Apply the knowledge of Lagrange's method of undetermined multipliers.	Chalk & Talk, Assignment Submission , Unit Test.
Week 4	Reduction formula, Curve tracing	Using some properties of the curve, one can identify the curve	Chalk & Talk, Assignment Submission , Unit Test.
Week 5	Double Integral, Change of order, Change of variables,	Find the solution of different types of mathematical models using knowledge about the possibilities and limitations of the different methods and tools.	Chalk & Talk, Assignment Submission , Unit Test.
Week 6	Triple integral, Application of multiple integrals	Apply the knowledge to find multiple integration	Chalk & Talk, Assignment

			Submission , Unit Test.
Week 7	Curvilinear coordinate system, Cartesian, Spherical and Cylindrical coordinate system	Apply the Knowledge of Curvilinear coordinate system.	Chalk & Talk, Assignment Submission , Unit Test.
Week 8	Vector Differentiation, Directional Derivative, Gradient of a Scalar Function and Conservative Field	Apply the Knowledge of vector differentiation for Directional derivative and Conservative field.	Chalk & Talk, Assignment Submission , Unit Test.
Week 9	Directional Derivative, Gradient of a Scalar Function and Conservative Field	Apply the Knowledge of vector differentiation for Directional derivative and Conservative field.	Chalk & Talk, Assignment Submission , Unit Test.
Week 10	Divergence and Curl, Related Properties of Gradient, Sums of Divergence and Curl	Apply the Knowledge of vector differentiation for Directional derivative and Conservative field.	Chalk & Talk, Assignment Submission , Unit Test.
Week 11	Vector Integration: Integration of a Vector Function of a Scalar Argument	Apply the Knowledge of Vector Integration and Line Integral.	Chalk & Talk, Assignment Submission , Unit Test.
Week 12	Line Integrals: Work Done, Potential, , Conservative Field and Area	Apply the Knowledge of Vector Integration and Line Integral.	Chalk & Talk, Assignment Submission , Unit Test.
Week 13	Introduction to Surface Integrals, Volume Integrals	Apply the Knowledge of Vector Integration and Line Integral.	Chalk & Talk, Assignment Submission , Unit Test.
Week 14	Green's Theorem in Plane, Stokes' Theorem	Apply the Knowledge of Vector Integration and Line Integral.	Chalk & Talk, Assignment Submission , Unit Test.
Week 15	Gauss Divergence Theorem	Apply the Knowledge of Vector Integration and Line Integral.	Chalk & Talk, Assignment Submission

🔵 📕 ज्ञानेन प्रकाशते जगत्



PROGRAM MAP for Bachelor of Engineering (Department of Mathematics, ISHLS)

Subject Mind Mapping

Sr. No.	Semester	Course Name	Compulsory/Open Elective
1	I	Calculus	Compulsory
2	II	Linear Algebra and Differential Equations	Compulsory
3	III	Probability , Statistics and Numerical methods	Compulsory
4	IV	Complex Analysis / Discrete Mathematics	Open elective
5	V	Finite Element method	Open elective
6	VI	Graph Theory	Open elective
7	VII	Optimization Techniques / Artificial neural network and soft computing	Open elective



Name of Institute: INDUS INSTITUTE OF TECHNOLOGY & ENGINEERING

Name of Faculty: Prof. Zankar Yadav

Course code: ME0112

Course name: Workshop Practice Pre-requisites: Basic Engineering Principles Credit points: 02 Offered Semester: 1st

Course Coordinator (weeks 14 - 16)

Full Name: Prof. Zankar Yadav
Department with sitting location: Ground Floor, Mechanical Workshop, IITE
Telephone: 8511225519
Email: zankarvyadav.me@indusuni.ac.in
Consultation times: 09.00 AM – 11.00 AM (Working Saturdays), 04:15 PM _ 05:00 PM (Monday to Friday)

Students will be contacted throughout the Session via Mail with important information relating to this Course.

Course Objectives

By participating in and understanding all facets of this Course a student will:

1. Recognize the importance of Workshop in today's technology and its impacts on market competition.

- 2. Understand basic of various methodologies to manufacture components/structure.
- 3. Understand different tools and equipments used in mechanical workshop.
- 4. Analyze the engineering and economical aspects of workshop practices.



Course Outcomes (CO)

After learning the course, the students should be able to:

- 1. Prepare basic joints used in carpentry.
- 2. Prepare edges for better joint for fitting.
- 3. Prepare better joint for welding.
- 4. Prepare various shapes and objects by using Carpentry, Fitting, and Welding.

Course Outline

1 Introduction to Mechanical Workshop

Introduction, Safety rules in Mechanical workshop, Safety slogans, Tools and Equipment used for safety in Mechanical workshop.

2 Workshop Layout

Introduction, Types of layouts, Rules and regulations for Workshop layouts and Preparation of actual layout.

3 Fitting Shop

Introduction, Classification and characteristics of Engineering Materials, Tools and Equipment for Fitting shop, Different operations, preparation of job in Fitting shop.

4 Carpentry Shop

Introduction, Wooden Materials, Tools and Equipment for Carpentry shop, Different operations, preparation of job in Carpentry shop.

Method of delivery

Face to face lectures, self study material, PPT, Web Resources

Study time

4 hours/Week

CO-PO Mapping (PO: Program Outcomes)

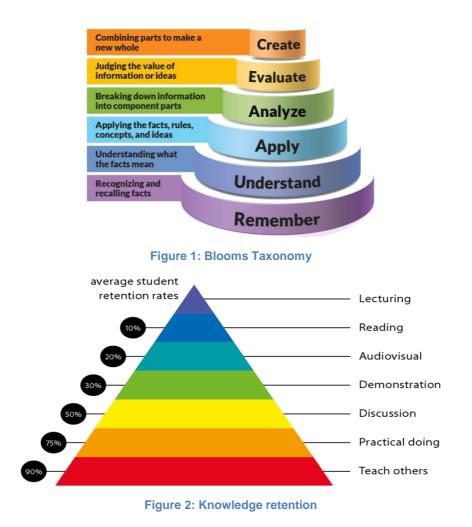
	P01	PO2	PO3	PO4	PO5	PO6	P07	PO8	PO9	PO10	P011	PO12
CO1	3	1	1	1	3	-	-	-	-	-	1	1



CO2	3	2	3	1	3	-	-	-	-	-	-	-
CO3	3	2	2	2	3	-	-	-	-	-	-	-

Blooms Taxonomyand Knowledge retention(For reference)

(Blooms taxonomy has been given for reference)



Graduate Qualities and Capabilities covered

(Qualities graduates harness crediting this Course)

General Graduate Qualities	Specific Department of Mechanical Engineering Graduate Capabilities
Informed Have a sound knowledge of an area of study or profession and understand its current issues, locally and internationally. Know how to apply this knowledge. Understand how an area	1 Professional knowledge, grounding & awareness



of study has developed and how it relates to other areas.	
Independent learners Engage with new ideas and ways of thinking and critically analyze issues. Seek to extend knowledge through ongoing research, enquiry and reflection. Find and evaluate information, using a variety of sources and technologies. Acknowledge the work and ideas of others.	2 Information literacy, gathering & processing
Problem solvers Take on challenges and opportunities. Apply creative, logical and critical thinking skills to respond effectively. Make and implement decisions. Be flexible, thorough, innovative and aim for high standards.	4 Problem solving skills
Effective communicators	5 Written communication
Articulate ideas and convey them effectively using a range of media. Work collaboratively	6 Oral communication
and engage with people in different settings. Recognize how culture can shape communication.	7 Teamwork
Responsible Understand how decisions can affect others and make ethically informed choices. Appreciate and respect diversity. Act with integrity as part of local, national, global and professional communities.	10 Sustainability, societal & environmental impact

Practical work:

- 1. To study about safety rules in Mechanical Workshop.
- 2. To study about Workshop layout and preparation of actual layout of Mechanical Workshop.
- 3. To study about Fitting shop and preparation of job.
- 4. To study about Carpentry shop and preparation of job.
- 5. To study about Tin Smithy shop and preparation of job.

Lecture/tutorial times

Attendance Requirements



The University norms states that it is the responsibility of students to attend all lectures, tutorials, seminars and practical work as stipulated in the course outline. Minimum attendance requirement as per university norms is compulsory for being eligible for semester examinations.

Details of referencing system to be used in written work

PPTs, Lecture Notes/E-book, Web-Resources.

Text books

1. Elements of Mechanical Engineering - Hajra Choudhury & others, Media Promoters 2010 .

2. The Elements of Workshop Technology - Volume I & II, S.K. Hajra choudhury, A.K.

Hajra Choudhury, Nirjhar Roy, 11th edition 2001 others, Media Promoters and Publishers, Mumbai.

ASSESSMENT GUIDELINES

Your final course mark will be calculated from the following:

CIE Practical 60 Marks Bifurcation			
10 Marks	Attendance & Lab Participation		
10 Marks	File work/ Journal write up – calculation etc. in lab. itself		
10 Marks	Fitting job		
10 Marks	Carpentry job		
10 Marks	Tin smithy job		
10 Marks	Question Answer		

SUPPLEMENTARY ASSESSMENT

Students who receive an overall mark less than 40% in internal component or less than 40% in the end semester will be considered for supplementary assessment in the respective components (i.e internal component or end semester) of semester concerned. Students must make themselves available during the supplementary examination period to take up the respective components (internal component or end semester) and need to obtain the required minimum 40% marks to clear the concerned components.

Practical Work Report/Laboratory Report:

A report on the practical work is due the subsequent week after completion of the class by each group.



Late Work

Late assignments will not be accepted without supporting documentation. Late submission of the reports will result in a deduction of -% of the maximum mark per calendar day

Format

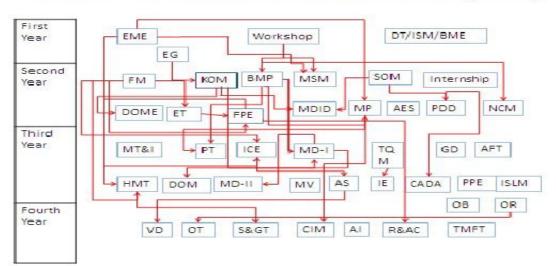
All assignments must be presented in a neat, legible format with all information sources correctly referenced. Assignment material handed in throughout the session that is not neat and legible will not be marked and will be returned to the student.

Retention of Written Work

Written assessment work will be retained by the Course coordinator/lecturer for two weeks after marking to be collected by the students.

University and Faculty Policies

Students should make themselves aware of the University and/or Faculty Policies regarding plagiarism, special consideration, supplementary examinations and other educational issues and student matters.



B.TECH MECHANICAL ENGINEERING (2019)